

Chemistry 344 -
Summary of Topics

I. Review of basic quantum mechanical principles

A. Scale - correspondence

- 1) the need for q.m. comes when classical mechanics fails for atomic scale systems - small mass & size
- 2) Early 20th century problems and solutions you've studied
 - a) Black body radiation
Planck - energy not continuous
 - b) Photoelectric effect
Einstein - light had particulate nature
• photon: $E = h\nu$
 - c) Atomic line spectra
Bohr - energy levels - light emitted by "jump" from one to other: $\Delta E = h\nu$
- angular momentum restricted: $L = n\hbar$
 - d) DeBroglie - wave nature of particles: $\lambda = h/p$
Proof - electron diffraction - Davisson & Germer
 - e) Spectroscopic linewidths, excited state life times
Heisenberg uncertainty principle
 $\Delta x \Delta p \geq \hbar/2$
 $\Delta E \Delta t \geq \hbar$

N.B. \rightarrow Opens era of modern quantum mechanics - must get idea straight.

3. Development of Quantum Mechanics

- generalized formulation contain all of the § 2 "patches" to classical

- a) Heisenberg - matrix mechanics - statistical
 - i) states = vectors
 - ii) observables = matrices
- b) Schrödinger - wave mechanics
 - i) states = wave functions
 - ii) observables = operators

Follows classical wave equation approach (in effort to contain duality)

4. Postulates - a set of ground rules from which all else can be derived.

Post. I. Particle described completely by wave function, $\psi(\underline{r}, t)$, which is:

- a) continuous, single valued, integrable square
- b) $\psi^* \psi d\tau \rightarrow$ probability interpretation
- c) Many (n) particles: $\psi(\underline{q}, t)$ - where $\underline{q} \rightarrow 3n$ positional coordinates

Post. II. Every observable corresponds to linear, hermetian operator.

- a) linear $\hat{\alpha}(f_1 + f_2) = \hat{\alpha}f_1 + \hat{\alpha}f_2$
- b) hermitian $\int \psi_n^* \hat{\alpha} \psi_m d\tau = \int \psi_m^* \hat{\alpha} \psi_n d\tau$
- c) Correspondence with observables
 - Constant: $a \rightarrow \hat{a}$
 - position variable: $x \rightarrow \hat{x}$
 - function of position: $f(x) \rightarrow \hat{f}(x)$
 - momentum: $p_x \rightarrow -i\hbar \frac{\partial}{\partial x}$

Post. III. If $\hat{\alpha}$ corresponds to an observable A, ψ_k is an eigenfunction of α with eigenvalue a_k and a set of identical systems are all in a state described by ψ_k ; then result of measuring A on any of the systems will be a_k every time.

Post. IV. If the systems in III are not in states described by eigenfunctions of α , then a measurement of A on each system may be different. The average value of the distribution is:

$$\langle \alpha \rangle = \frac{\int \psi_s^* \hat{\alpha} \psi_s d\tau}{\int \psi_s^* \psi_s d\tau}$$

where ψ_s is the wave function that describes each system.

Post. V. Time evolution of $\psi(\underline{q}, t)$ is given by:

$$i\hbar \frac{\partial}{\partial t} \Psi(\underline{q}, t) = \mathcal{H} \Psi(\underline{q}, t)$$

where \mathcal{H} is the hamiltonian (total energy) operator.

$$\text{--- if } \mathcal{H} \neq \mathcal{H}(t) \quad \Psi(\underline{q}, t) = \Psi(\underline{q}) e^{-i E t / \hbar}$$

5. Resultant properties of \hat{Q} .m. fct. & op. (fct. = abbreviation for function)

a. Orthonormality of eigenfunction:

normalized: $\int \psi_k^* \psi_k d\tau = 1$

orthogonal: $\int \psi_k^* \psi_{k'} d\tau = \delta_{kk'}$

b. Completeness: set of n fct ψ_i all orthonormal and spanning n -dimensional space is complete
 - any fct ϕ can be expanded in complete set

$$\phi = \sum_i c_i \psi_i \text{ where } c_i = \int \psi_i^* \phi d\tau$$

c. Average value: $\langle \alpha \rangle = \frac{\int \phi^* \hat{\alpha} \phi d\tau}{\int \phi^* \phi d\tau}$

Contribution of each eigen value a_k weighted by "projection" of ϕ upon "basis set" ψ_k 's

d. Commutation: in general $\hat{\alpha}\hat{\beta} \neq \hat{\beta}\hat{\alpha}$

if $[\hat{\alpha}, \hat{\beta}] = \hat{\alpha}\hat{\beta} - \hat{\beta}\hat{\alpha} = 0 \rightarrow$ commute

(remember these are operators)

If two operators commute they will have simultaneous eigenfunctions: if $[A, B] = 0$

e. Relationship to Uncertainty Principle, consider $\hat{\alpha}, \hat{\beta}$
 - if commute - same eigenfct; can measure observables of both equally accurate

- if do not commute - then system can only be in eigenfct of one or other, i.e. if one precise, other unknown.

B. Model systems

1. Particle in a box

a) $\mathcal{H} = -\hbar^2/2m \partial^2/\partial x^2$

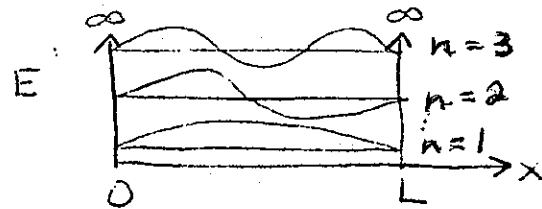
b) B.C. $\Psi(L) = \Psi(0) = 0$
 \hookrightarrow Boundary Condition

c) $\Psi = A \sin \alpha x$ - eigen function

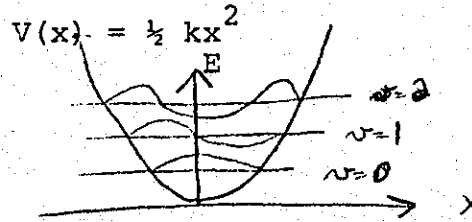
$\alpha = n\pi/L$ - from B.C.

$A = \sqrt{2/L}$ - from normalization

d) $E_n = \frac{\hbar^2}{2m} \frac{n^2 \pi^2}{L^2}$ - quantized energy



2. Harmonic oscillator



a) $\mathcal{H} = -\frac{\hbar^2}{2m} \nabla^2 + \frac{1}{2} kx^2$

b) B.C. $\Rightarrow \psi(\pm \infty) = 0$

c) $\psi_v = N_v H_v(y) \exp\left(-\frac{y^2}{2}\right) \rightarrow$ solution

where $y = \sqrt{\beta} x$, $\beta = \frac{\sqrt{km}}{\hbar}$; N_v - normalization factor

$H_v(y)$ - hermite polynomial

$$H_v(y) = (-1)^v e^{y^2} \frac{d^v}{dy^v} (e^{-y^2})$$

d) $E_v = (v + 1/2) \hbar \omega$; $\omega = 2\pi\nu = \sqrt{k/m}$ = angular frequency of vibration

Application - Vibrational spectroscopy, isotope effects

3. Rigid Rotor - 2 particle system, change of variable

a) $\mathcal{H} = -\frac{\hbar^2}{2m_1} \nabla_1^2 - \frac{\hbar^2}{2m_2} \nabla_2^2$ (6-dimension)

b) Center of Mass coordinates allow reduction to 3-D:

$$\mathcal{H}_r = \frac{\hbar^2}{2\mu} \nabla_r^2$$

relative coord

$$x = x_1 - x_2$$

$$y = x_1 - y_2$$

$$z = z_1 - z_2$$

$$\mu = \frac{m_1 m_2}{m_1 + m_2}$$

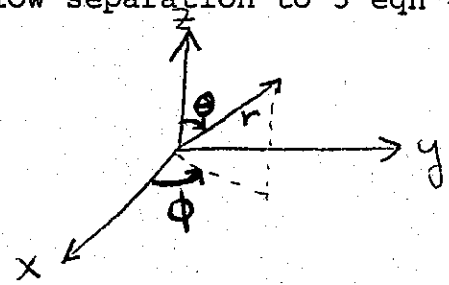
$$I = \mu r^2$$
 -moment of inertia

c) Spherical coordinates allow separation to 3 eqn :

$$x = r \cos \phi \sin \theta$$

$$y = r \sin \phi \sin \theta$$

$$z = r \cos \theta$$



Since r - constant (rigid) - 2 equations:

$$\frac{d^2 \phi}{d\phi^2} = -m^2 \phi$$

$$\frac{d^2 \psi}{d\theta^2} + \frac{\cos \theta}{\sin \theta} \frac{d\psi}{d\theta} + \left(\frac{2IE}{\hbar^2} - \frac{m^2}{\sin^2 \theta} \right) \psi = 0$$

d) Soln: $\psi_m(\phi) = e^{im\phi}$ $m = 0, \pm 1, \pm 2 \dots$

$\psi_{lm}(\theta) = P_l^{|m|}(\cos \theta)$ - assoc. Legendre

} not normalized

$$P_l^{|m|}(x) = (1-x^2)^{m/2} \frac{d^{|m|}}{dx^{|m|}} P_l(x)$$

$$P_l(x) = \frac{1}{2^l l!} \frac{d^l}{dx^l} (x^2-1)^l$$

limits

$l = 0, 1, 2, 3 \dots$ pos. integers

$|m| \leq l$ and $m = \text{integer} \Rightarrow m = 0, \pm 1, \pm 2 \dots \pm l$

e) $E_l = \frac{l(l+1)\hbar^2}{2I}$

i) each level $2l + 1$ degenerate, $E_l \neq f(m)$

ii) ang momentum limited

Application - microwave spectroscopy,

f) Spherical Harmonics $Y_{lm} = N_{lm} \psi_{lm} \Phi_m$

(here N_{lm} = normalization constant)

- eigen function of ang. mom: $\vec{L} = \vec{r} \times \vec{p}$

$$\hat{L}_z Y_{\ell m} = m\hbar Y_{\ell m}$$

$$\hat{L}^2 Y_{\ell m} = \ell(\ell+1)\hbar^2 Y_{\ell m}$$

4) Hydrogen Atom - 2 particles, central potential

a. $\mathcal{H} = -\frac{\hbar^2}{2m_1} \nabla_1^2 - \frac{\hbar^2}{2m_2} \nabla_2^2 - \frac{Ze^2}{r}$

b. Separate variables as for rigid rotor -

note: $V = V(r)$ only

c. θ, ϕ sol'n same as rotor: $Y_{\ell m}(\theta, \phi)$

d. r: $\frac{d^2 R}{dr^2} + \frac{2}{r} \frac{dR}{dr} + \left(\frac{2mE}{\hbar^2} + \frac{2mZe^2}{\hbar^2 r} - \frac{\ell(\ell+1)}{r^2} \right) R = 0$

sol'n: $R_{n\ell} = N x^\ell e^{-x/2} L_{n+\ell}^{2\ell+1}(x)$

where $x = \frac{2Zr}{na_0}$

$L_{n+\ell}^{2\ell+1}$ - LaGuerne polynomial

$n = 1, 2, 3, \dots$ pos. integers

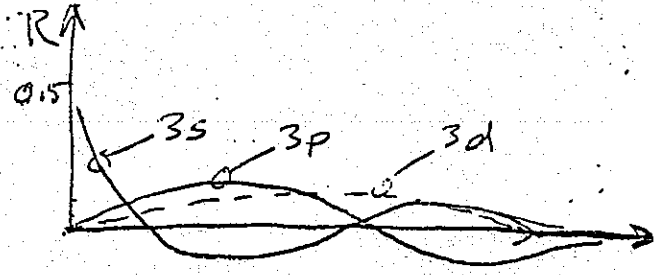
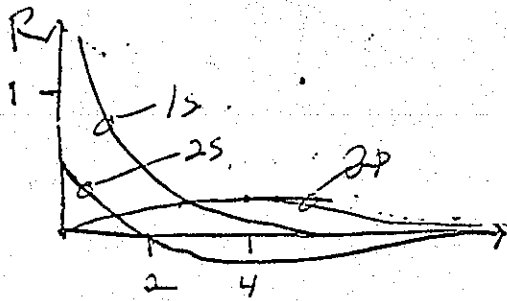
e. total solution:

$$\Psi_{n\ell m} = R_{n\ell}(r) Y_{\ell m}(\theta, \phi)$$

$$E_n = -\frac{m e^4 z^2}{2 n^2 \hbar^2}$$

Hydrogen atom fundamental to all of quantum chemistry
 more difficult problems typically use H-atom w/f as a
basis set

Hence important to understand shapes and forms of
 solutions:



note: nodes # = $n - l - 1$

