

Chemistry 542
Fall, 2002
Problem Set 7
Due Wednesday, Oct. 30

Read chapter 6, sections 1-4.

1. Show that the associated Legendre equation is a Sturm-Liouville equation and determine the weighting factor, $r(x)$.
2. Use ladder operators to derive $Y_{2,2}$, $Y_{2,1}$, and $Y_{2,0}$. Don't worry about the normalization.
3. Calculate the rotational population distribution for HCl and H_2 at 300 K. Present your results in the form of a table of P_J vs J . What is the most probable value of J for each molecule? For HCl it is OK to use the equation that we derived in class, but for H_2 the formula is inaccurate. Compare the exact result for with the values predicted by the formula.
4. The alignment of a rotor, A , is defined as the expectation value of $\cos^2 \theta$.
 - A. Calculate A for a molecule in state $J=0$.
 - B. Calculate A for $J = 2$, $M = 2$ by direct evaluation of the integrals.
 - C. Use a recursion relation to derive a general expression for the alignment of any J, M state. Show that your equation works for $J=1$, $M= 1$ and 0 , which were obtained in class.
 - D. Show that the average over all M gives $A = 1/3$ for any J . Assume that all M states are equally populated. Hint: $\sum_{M=-J}^J M^2 = J(J+1)(2J+1)/3$.
5. Prove that the recurrence time for a rotational wave packet is $1/2Bc$.

Also do Levine 6.4 and 6.6.