

Chemistry 542
 Fall, 2002
 Problem Set 5
 Due Monday, October 14

Read chapter 4, sections 1-3.

1. What is the probability of finding a harmonic oscillator in state $v=1$ at a distance greater than the classical turning point? First determine the turning point in reduced units and then evaluate the probability of y being greater than this value.
2. Use the recursion relation to determine the expectation value of x^4 for a harmonic oscillator in state v . If you pay close attention to the lecture, you can save yourself a lot of work.
3. Analyze the vibrational properties of the HCl molecule, using the table from Herzberg. The data are given in wave numbers.
 - A. Calculate the force constant for the ground electronic state.
 - B. Calculate the vibrational energies the $v=0$ and 1 states, treating the molecule as anharmonic.
 - C. Calculate the wavelength of the 0-1 transition.
 - D. Calculate the zero point displacement (i.e., the maximum stretch allowed classically) for $v=0$, assuming a harmonic wave function.
 - E. Repeat parts A-D for the V state of HCl.

State	T_e	ω_e	$\omega_e x_e$	B_e	α_e	D_e (10^{-4} cm^{-1})	r_e (\AA)
$^1\text{H}^{35}\text{Cl}$ (continued)							
x $^1\Sigma^+$	0	2990.946_3^x	52.8186^y	10.59341_6^{xz}	$0.30718_1^{a'}$	$5.3194^{zb'}$	$1.27455_2^{c'}$
v $^1\Sigma^+$	77293.0	877.16^z	16.04^r	2.727	-0.026	1.02^r	2.512
Continuous emission spectrum with maximum at 38900 cm^{-1} .							

4. Fill in the following details of the harmonic oscillator problem:
 - A Show that the classical kinetic energy of a harmonic oscillator is given by

$$T = \frac{1}{2} M \dot{R}^2 + \frac{1}{2} \mu \dot{r}^2$$

- B Demonstrate that $\psi(y) \approx e^{-y^2/2}$ is an approximate solution derivation of the Schrodinger equation at large y .

C Using the transformation $\psi(y) = \phi(y)e^{-\frac{1}{2}y^2}$, show that the Schrodinger equation becomes

$$\phi'' - 2y\phi' + (\alpha^2\beta - 1)\phi = 0.$$

Levine, problems 11, 16, 25, 26, 27.